



**CURRICULUM**  
**Subject: Science( 086)**  
**Sesion: 2023-24**  
**CLASS – IX**

EVALUATION SCHEME		
Theory		
Unit No.	UNITS	Marks
I	Matter-Its Nature and Behaviour	25
II	Organization in the Living World	22
III	Motion, Force and laws of motion	27
IV	Food; Food Production	06
Total		80
Internal Assessment		20
Grand Total		100

**Syllabus for Purpose of Examination 2023-24**  
**CLASS IX**

Chapter No/ Month.	Name of chapter/Learning outcomes	Practical and Competency Skill Based Activities/ Experiential Learning	Skills	Assessments
<b>Biology:</b> <b>Chapter: 5</b> <b>(Feb- March - April)</b>	<b>The Fundamental Unit of Life</b> <b>Cell - Basic Unit of life:</b> <b>Learning outcomes:</b> Student will be able to: <ul style="list-style-type: none"> <li>Understand:- cell as a basic unit of life</li> <li>Differentiate:- prokaryotic and eukaryotic cells, multicellular organisms;</li> <li>Describe:- cell membrane and cell wall, cell organelles and cell inclusions; chloroplast, mitochondria, vacuoles, endoplasmic reticulum, Golgi apparatus; nucleus, chromosomes - basic structure, number.</li> </ul>	<ul style="list-style-type: none"> <li>Preparation of stained temporary mounts of (a) onion peel, (b) human cheek cells &amp; to record observations and draw their labeled diagrams.</li> </ul>	Knowledge, Understanding, Application, Analysis and Evaluation	Oral Test/ Class test/ Quizzes / lab activity
<b>Chapter- 2</b> <b>(Feb- March- April)</b>	<b>IS MATTER AROUND US PURE</b> <b>Learning outcomes:</b> Student will be able to: <ul style="list-style-type: none"> <li>Differentiate between a pure substance and an impure substance present around them</li> <li>Tabulate matter and its types and further complex branching of matter</li> <li>Classify the elements based upon their physical and chemical properties</li> <li>Explain the properties of metals and nonmetals and their applications in daily life.</li> </ul>	<ul style="list-style-type: none"> <li>Preparation of: a) a true solution of common salt,sugar and alum b) a suspension of soil, chalk powder and fine sand in water c) a colloidal solution of starch in water and egg albumin/milk in water and distinguish between these on the basis of</li> </ul>	Knowledge, Understanding, Application, Analysis and Evaluation	Oral Test/ Class test/ Quizzes on google forms/ lab activity

	<ul style="list-style-type: none"> <li>Differentiate between aqueous and non aqueous solutions</li> <li>Calculate the concentrations of the solutions by applying appropriate formulas.</li> <li>categorise saturated unsaturated and supersaturated solution</li> <li>Describe the Properties of suspensions</li> <li>Define a colloidal solution w.r.t. the dispersed phase and dispersion medium</li> <li>Categorise the types of colloidal systems</li> </ul>	<ul style="list-style-type: none"> <li>transparency</li> <li>filtration criterion</li> <li>stability</li> </ul> <p>Unit-I: (Chapter -2) 2. Preparation of</p> <p>a) A mixture</p> <p>b) A compound using iron filings and sulphur powder and distinguishing between these on the basis of:</p> <p>i. appearance, i.e., homogeneity and heterogeneity</p> <p>ii. behavior towards a magnet</p> <p>iii. behavior towards carbon disulphide as a solvent</p> <p>iv. effect of heat</p>		
<b>Chemistry:</b> <b>Chapter - 1</b> <b>(May-June)</b>	<p><b>MATTER IN OUR SURROUNDINGS:</b></p> <p><b>Learning outcomes:</b></p> <p>Students will be able to</p> <ul style="list-style-type: none"> <li>Define matter</li> <li>Categorise matter into solid, liquid and gas</li> <li>Understand the characteristics - shape, volume, density</li> <li>Identify the change of state-melting (absorption of heat), freezing, evaporation (cooling by evaporation), condensation, sublimation.</li> </ul>	<ul style="list-style-type: none"> <li>Perform the following reactions and classify them as physical or chemical changes</li> <li>a) Iron with copper sulphate solution in water</li> <li>b) Burning of magnesium ribbon in air</li> <li>c) Zinc with dilute sulphuric acid</li> <li>d) Heating of copper sulphate crystals</li> <li>e) Sodium sulphate with barium chloride in the form of their solutions in water.</li> </ul>		
<b>Physics:</b> <b>Chapter -8</b> <b>(Feb-March - April)</b>	<p><b>Motion</b></p> <p><b>Learning outcomes</b></p> <p>Student will be able to:</p> <ul style="list-style-type: none"> <li>Define: Distance and displacement, velocity; uniform and non-uniform motion along a straight line; acceleration,</li> <li>Draw: distance-time and velocity-time graphs for uniform motion and uniformly accelerated motion.</li> <li>Derive: equations of motion by graphical method;</li> <li>Explain: elementary idea of uniform circular motion.</li> </ul>	<ul style="list-style-type: none"> <li>Inter- class quiz And numerical based on different terms ( Distance, Displacement, speed, velocity, acceleration)</li> </ul>	Knowledge, Understanding, Application, Analysis and Evaluation	Oral Test/ Class test/ Quizzes / lab activity.
<b>Biology:</b> <b>Chapter-6</b> <b>( May - June - July)</b>	<p><b>Tissues, Organs, Organ System, Organism:</b></p> <p><b>Learning outcomes:</b></p> <p>Student will be able to:</p> <ul style="list-style-type: none"> <li><b>Describe:-</b> Structure and functions of animal and plant tissues (only four types of tissues in animals;</li> <li>Differentiate between:- Meristematic and Permanent tissues in plants).</li> </ul>	<ul style="list-style-type: none"> <li>Identification of Parenchyma, Collenchyma and Sclerenchyma tissues in plants, striped, smooth and cardiac muscle fibers and nerve cells in animals, from prepared slides. Draw their labeled diagrams.</li> </ul>	Knowledge, Understanding, Application, Analysis and Evaluation	Oral Test/ Class test/ Quizzes / lab activity.
<b>Physics:</b> <b>( May - June - July)</b>	<p><b>Force and Laws of Motion</b></p> <p><b>Learning outcomes</b></p> <p>Student will be able to:</p> <ul style="list-style-type: none"> <li>Define: Force and Newton's laws: Force and Motion, Newton's Laws of Motion, Action and Reaction forces, Inertia of a body.</li> <li>Explain: Inertia and mass, Momentum, Force and Acceleration.</li> <li>Define: Elementary idea of conservation of Momentum.</li> </ul>	<ul style="list-style-type: none"> <li>Inter- class quiz And numerical based on motion, momentum, and conservation of momentum.</li> </ul>	Knowledge, Understanding, Application, Analysis and Evaluation	Oral Test/ Class test/ Quizzes / lab activity

<b>Chemistry:</b> <b>Chapter- 3</b> <b>(July - Aug)</b>	<b>ATOMS AND MOLECULES</b> <b>Learning outcomes:</b> Students will be able to: <ul style="list-style-type: none"> <li>● Explain the law of conservation of mass and law of constant composition</li> <li>● Tell the experiment to verify the law of conservation of mass</li> <li>● solve the numericals based upon the law of conservation of mass and law of constant composition</li> <li>● Give the postulates of atomic theory by Dalton.</li> <li>● Calculate the limitations or drawbacks of Dalton's atomic theory</li> <li>● Calculate and define the atomic mass and relative atomic mass</li> <li>● Differentiate between Molecules of an element and molecules of a compound</li> <li>● Calculate the molecular mass of different Compounds</li> <li>● Write the chemical formula of compounds</li> <li>● Have a clear vision about gram atomic mass and gram molecular mass</li> <li>● Introduction of moles concept.</li> </ul>	<ul style="list-style-type: none"> <li>● Solving numericals based upon the mole concept.</li> <li>● Verification of the law of conservation of mass in a chemical reaction</li> </ul>	Knowledge, Understanding, Application, Analysis and Evaluation	Oral Test/ Class test/ Quizzes on google forms/ lab activity
<b>Physics:</b> <b>Chapter-10</b> <b>(July-Aug)</b>	<b>Gravitation</b> <b>Learning outcomes</b> Student will be able to: <ul style="list-style-type: none"> <li>● Explain: Gravitation; Universal Law of Gravitation.</li> <li>● Define: Force of Gravitation of the earth (gravity).</li> <li>● Explain: Acceleration due to Gravity.</li> <li>● Differentiate: Mass and Weight; Free fall.</li> </ul>	<ul style="list-style-type: none"> <li>● Determination of the density of solid (denser than water) by using a spring balance and a measuring cylinder.</li> <li>● Establishing the relation between the loss in weight of a solid when fully immersed in a) Tap water]</li> <li>● Strongly salty water with the weight of water displaced by it by taking at least two different solids</li> <li>● Determination of the melting point of ice and the boiling point of water.</li> </ul>	Knowledge, Understanding, Application, Analysis and Evaluation	Oral Test/ Class test/ Quizzes / lab activity
<b>Biology:</b> <b>Chapter- 13</b> <b>(Aug.-Sept - Oct)</b>	<ul style="list-style-type: none"> <li>● <b>Food Production</b></li> </ul> <b>Students will be able to:</b> <ul style="list-style-type: none"> <li>● <b>Understand:</b> Plant and animal breeding and selection for quality improvement and management; Describe: Use of fertilizers and manures; Explain: Protection from pests and diseases; Organic farming.</li> </ul>	<ul style="list-style-type: none"> <li>● Inter - class quiz on different food resources ( Plants and Animals)</li> </ul>	Knowledge, Understanding, Application, Analysis and Evaluation	Oral Test/ Class test/ Quizzes / lab activity

<b>Chemistry:</b> <b>Chapter - 4</b> <b>(September - October)</b>	<b>STRUCTURE OF ATOM</b> <b>Learning outcomes:</b> Students will be able to: <ul style="list-style-type: none"> <li>● Explain: the discovery of electrons or study of cathode rays.</li> <li>● Analyze the properties of cathode rays</li> <li>● Define electrons and write the charge and mass on electron</li> <li>● Describe the origin and production of anode rays</li> <li>● Analyze the properties of anode rays</li> <li>● Describe Thomson model of atom</li> <li>● Describe Rutherford model of atom</li> <li>● Detailed explanation of Bohr's model atom</li> <li>● Relate the atomic number and mass number</li> <li>● Calculate the number of electrons protons and neutrons from atomic number and mass number</li> <li>● Calculate the valence electrons and valency of an element</li> <li>● Give examples of isotopes and define them</li> <li>● Describe the applications of isotopes</li> <li>● Define isobars</li> </ul>		Knowledge, Understanding, Application, Analysis and Evaluation	Oral Test/ Class test/ Quizzes / lab activity
<b>Physics:</b> <b>chapter-11</b> <b>(September - October)</b>	<b>Work, energy and power:</b> <b>Learning outcomes</b> Student will be able to: <ul style="list-style-type: none"> <li>● Define: Work done by a Force.</li> <li>● Explain: Energy, power.</li> <li>● Define: Kinetic and Potential energy; Law of conservation of energy.</li> </ul>	<ul style="list-style-type: none"> <li>● Numerical based on work power and energy</li> </ul>	Knowledge, Understanding, Application, Analysis and Evaluation	Oral Test/ Class test/ Quizzes / lab activity
<b>Physics</b> <b>Chapter-12</b> <b>November</b>	<b>Sound</b> Students will be able to: <ul style="list-style-type: none"> <li>● Understand: Nature of sound and its propagation in various media</li> <li>● Describe: Echo and Sonar.</li> <li>● Explain: Structure of the Human Ear (Auditory aspect only)</li> </ul>	<ul style="list-style-type: none"> <li>● Verification of the Laws of reflection of sound.</li> <li>● Determination of the speed of a pulse propagated through a stretched string/slinky (helical spring).</li> <li>● <b>Competency based activity</b> To analyze national anthem on the basis of pitch and amplitude.</li> </ul>	Knowledge, Understanding, Application, Analysis and Evaluation	Oral Test/ Class test/ Quizzes / lab activity
<b>Biology:</b> <b>Chapter- 14</b> <b>October</b>	<b>Our Environment</b> Natural Resources Students will be able to: <ul style="list-style-type: none"> <li>● Describe:- Physical resources: Air, Water, Soil. Air for respiration, for combustion, for moderating temperatures; movements of air and its role in bringing rains across India.</li> <li>● Analyze:- Air, water and soil pollution (brief introduction).Holes in ozone layer and the probable damages.</li> </ul>	<b>Portfolio/File presentation</b> <b>Based on natural resources</b>	Knowledge, Understanding, Application, Analysis and Evaluation	Oral Test/ Class test/ Quizzes / lab activity